## REINVESTIGATION OF THE JUTZ SYNTHESIS OF BENZO[e] PYRENE AND BENZO[a] PYRENE DERIVATIVES FROM BENZANTHRENE

Hongmee Lee and Ronald G. Harvey

Ben May Laboratory, University of Chicago Chicago, Illinois 60637

Abstract: Base-catalyzed reactions of benzanthrene with "vinamidinium salts" (2a-c) followed by thermal electrocyclic ring closure are regiospecific affording only benzo[e] pyrene derivatives, contrary to previous claims.

Reaction of benzanthrene (1) with "vinamidinium salts" (2) is reported by Jutz<sup>1</sup> to furnish condensation products which undergo thermal electrocyclic ring closure with elimination of amine to provide the corresponding derivatives of benzo[e] pyrene (3) and benzo[a] pyrene (4). Reaction of the unsubstituted salt 2a is reported to furnish the isomeric benzopyrenes 3a and 4a in 2:1 ratio. 1

Careful reinvestigation of the Jutz synthesis with 2a afforded, however, only 3a with no detectable trace of 4a by NMR<sup>2</sup> or TLC<sup>3</sup> analysis of the crude or purified product. Analogous reactions of the methyl- and ethoxy-substituted salts 2b and 2c gave similar results. The related salt  $(CH_3)_2$  N=CH-CH=C(CH<sub>3</sub>)N(CH<sub>3</sub>)<sub>2</sub> ClO<sub>4</sub> (5), bearing the methyl group on the terminal carbon atom, underwent similar reaction to provide 3-methylbenzo[e] pyrene<sup>5</sup> as the sole detectable polyarene product.<sup>6</sup>

Reactions were conducted essentially according to the reported procedures. Synthesis of 2-ethoxybenzo[e] pyrene  $(\underline{6})$ , not previously described, was carried out by heating a solution of  $\underline{1}$  (320 mg, 1.5 mmol), NaOMe (89 mg, 1.65 mmol) and  $\underline{2c}^7$  (407 mg, 1.5 mmol) in pyridine (10 ml) at 100° for 5 hr under N<sub>2</sub>. The pyridine was then replaced by quinoline (5 ml) and the solution was heated at reflux overnight. Conventional workup and passage through a column of Florisil eluted with benzene-hexane (1:1) gave  $\underline{6}$  (142 mg, 32%) as a white solid, mp 120-122°C: NMR  $\underline{6}$  1.5 (t,3,CH<sub>3</sub>), 4.2 (q,2,CH<sub>2</sub>), 7.4-8.1 (m,7, aromatic), 8.22 (d,1,H<sub>8</sub>), 8.5-8.8 (m,3,H<sub>1,9,12</sub>). The four bay region protons (H<sub>1,8,9,12</sub>) which appear at characteristic low field confirm the benzo[e] pyrene structural assignment.

In the prior studies only a single isomeric benzopyrene derivative was isolated and characterized from each experiment. The second isomer was inferred to be present on the basis of a color test with  ${\rm H_2SO}_4$  and the broad melting range of the products. In our hands, the color test with authentic benzo[e] pyrene and benzo[a] pyrene showed no clearly distinctive difference. The compounds previously identified as benzo[a] pyrene and 2-methylbenzo[a] pyrene are now reassigned as benzo[e] pyrene and 2-methylbenzo[e] pyrene, respectively.

We conclude that the base-catalyzed reactions of benzanthrene with the salts 2a-c and 5 followed by thermal electrocyclic ring closure are essentially regiospecific, affording only the corresponding benzo[e] pyrene derivatives. These reactions provide convenient synthetic access to benzo[e] pyrene derivatives which are otherwise difficulty obtainable.

## REFERENCES AND FOOTNOTES

- J. C. Jutz, <u>Topics in Current Chem.</u>, 73, 127 (1978); J. C. Jutz, R. Kirchlechner, and H.-J. Seidel, Chem. Ber, 102, 2301 (1969).
- 2. The high resolution 270 MHz NMR spectra of the isolated products of the reactions of 2a and 2b exhibited chemical shifts and splitting patterns interpretable only in terms of the benzo[e] pyrene isomers 3a and 3b. NMR of 3a (CDCl<sub>3</sub>)  $\delta$  7.69 (m,2,H<sub>10,11</sub>,J<sub>9,10</sub>=J<sub>11,12</sub>= 6 Hz,J<sub>9,11</sub>=J<sub>10,12</sub>= 3 Hz), 7.97 (m,2,H<sub>2,7</sub>,J<sub>1,2</sub>=J<sub>2,3</sub>=J<sub>6,7</sub>=J<sub>7,8</sub>= 7.8 Hz), 7.99 (s,2,H<sub>4,5</sub>), 8.13 (m,2,H<sub>3,6</sub>,J<sub>1,3</sub>=J<sub>6,8</sub>= 1.5 Hz), 8.78 (m,2,H<sub>9,12</sub>), 8.83 (m,2,H<sub>1,8</sub>). NMR of 3b (CDCl<sub>3</sub>)  $\delta$  2.82 (s,3,CH<sub>3</sub>), 7.72 (m,2,H<sub>10,11</sub>,J<sub>9,10</sub>=J<sub>11,12</sub>= 6.3 Hz), J<sub>9,10</sub>=J<sub>10,12</sub>=2.8 Hz), 7.95 (d,1,H<sub>4</sub> or 5,J<sub>4,5</sub>= 9.5 Hz), 7.96 (t,1,H<sub>7</sub>,J<sub>6,7</sub>=J<sub>7,8</sub>= 9 Hz), 7.98 (d,1,H<sub>4</sub> or 5), 8.14 (m,1,H<sub>6</sub>), 8.69 (s,1,H<sub>1</sub>), 8.82 (m,2,H<sub>9,12</sub>), 8.86 (m,1,H<sub>8</sub>).
- 3. TLC analysis of the products of the reaction of 2a on 2,4,7-trinitrofluorenone-impregnated silica gel<sup>4</sup> revealed only the characteristic orange-brown spot of 3a with no indication of the distinctive black spot of 4a.
- 4. R. G. Harvey and M. Halonen, <u>J. Chromatog.</u>, 25, 294 (1966).
- 5. The NMR spectrum of 3-methylbenzo[e] pyrene matched that of an authentic sample.
- 6. H. Lee, N. Shyamasundar, P. P. Fu, and R. G. Harvey, J. Org. Chem., submitted.
- 7. A. Arnold, Coll. Czech. Chem. Commun., 38 1168 (1973).
- 8. Also detected as products of these reactions were benzanthrone ( $\sim 30\%$ ) and a volatile compound, possibly methylquinoline, which exhibited a methyl peak as a singlet at  $\delta$  2.5 in the NMR and a major peak at m/e = 142 in the mass spectrum.